

# *A Comprehensive Review on l-Asparaginase and Its Applications*

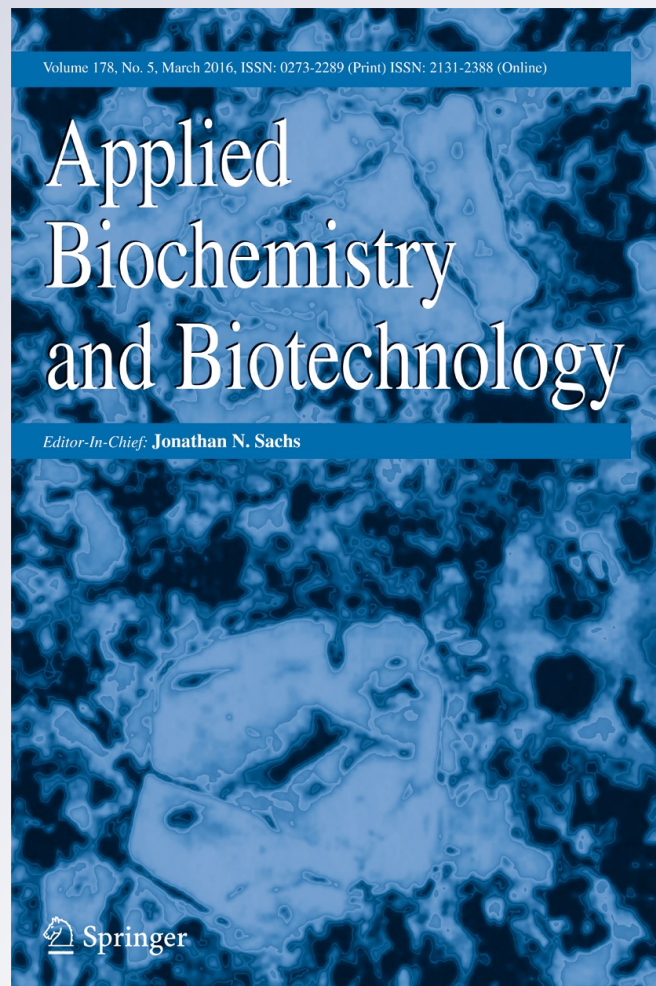
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# A Comprehensive Review on L-Asparaginase and Its Applications

Tahira Batool<sup>1</sup> · Essam A. Makky<sup>1</sup> · Muna Jalal<sup>1,2</sup> · Mashitah M. Yusoff<sup>1</sup>

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**Abstract** L-asparaginase (LA) catalyzes the degradation of asparagine, an essential amino acid for leukemic cells, into ammonia and aspartate. Owing to its ability to inhibit protein biosynthesis in lymphoblasts, LA is used to treat acute lymphoblastic leukemia (ALL). Different isozymes of this enzyme have been isolated from a wide range of organisms, including plants and terrestrial and marine microorganisms. Pieces of information about the three-dimensional structure of L-asparaginase from *Escherichia coli* and *Erwinia* sp. have identified residues that are essential for catalytic activity. This review catalogues the major sources of L-asparaginase, the methods of its production through the solid state (SSF) and submerged (SmF) fermentation, purification, and characterization as well as its biological roles. In the same breath, this article explores both the past and present applications of this important enzyme and discusses its future prospects.

**Keywords** L-asparaginase · Lymphoblasts · Acute lymphoblastic leukemia · Solid state fermentation · Submerged fermentation

## Introduction

All enzymes are proteins (and in some cases RNA molecules) with varied functions. They catalyze reactions used in normal development, help in the maintenance of the cell, and work as defense against diseases. Enzymes can operate intracellularly, extracellularly, or even on the surface of a cell membrane [1]. Several hundred different enzymes have been identified, and many of them have been characterized to a considerable degree. Quite a few of these enzymes

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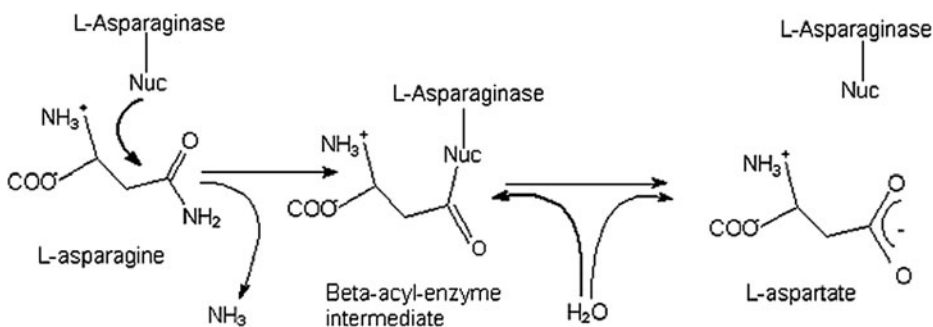
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have future of particular interest and command the attention of loyal bands of investigators. However, only a relatively small number have moved to the center stage and turned into the objects of extensive investigations. According to Hammes, [2], such enzymes come up with special experimental advantages. One of such enzymes is L-asparaginase. L-Asparaginase (L-asparagine amidohydrolase, EC 3.5.1.1) has been the focus of several papers in the most recent three decades. Although it does have certain features of interest, it must be admitted that L-asparaginase is not especially remarkable as an enzyme. What is unusual about asparaginase and constitutes the feature remarkable for the considerable attention it has attracted is its antineoplastic activity. L-asparaginase also known as aminohydrolase pertains to the amidase group of enzymes and can break down amino acid L-asparagine into aspartate and ammonia [3] (Fig. 1).

L-asparaginase is the first therapeutic enzyme with antineoplastic properties that has been studied broadly by researchers and scientists far and wide. L-asparaginase was first observed by Lang in 1904 [4]. Research on physiological capabilities of L-asparaginase was under way for more than half century, and a real leap forward was accomplished in 1922 when Clementi [5] revealed the presence of L-asparaginase in blood serum of guinea pig. Kidd [6] in 1953 did a progression of experiments to prove the ability of guinea pig serum as tumor inhibitor. He carried out diverse sets of analysis where he concocted the findings that two sorts of subcutaneously executed lymphosarcoma in mice failed to proliferate once animals were given injections of guinea pig serum, while untreated controlled mice died because of carcinomas. In the second set of experiments, two different types of lymphomas namely mammary carcinoma and fibrosarcoma of mice failed to degenerate when treated with guinea pig serum. In 1961, yet another breakthrough was achieved when Broome [7] demonstrated L-asparaginase as an antitumor agent in guinea pig serum with substrate specificity. Later on, some other substrate-specific L-asparaginases were found to inhibit tumors [8–10]. McCoy through his in vitro experiments proved the centrality of amino acid asparagine for Walker carcinoma 256 [11]. Altenbern in 1954 and Broome in 1965 reported the antitumor activity of L-asparaginase in bacteria and yeast, respectively [12, 13]. L-asparaginase is found in the serum of guinea pig and rodents but is absent in humans [14]. Tumors in mice, dogs, and rats can be regressed by the asparaginase enzyme purified from guinea pig serum, *Escherichia coli*, and to a lesser degree by the chicken liver [15–17]. Two isozymes of L-asparaginase, namely type I and type II, have been identified by Ohnuma in 1967 [18]. Both type I and type II asparaginases are characterized by enzymatic activity for both L-asparagine and L-glutamine. However, type II asparaginase displays higher specific action against L-asparagine. Type II asparaginase precisely shows antitumor activity and is utilized as chemotherapeutics in ALL [19].



**Fig. 1** Schematic illustration of the reaction mechanism of L-asparaginase

Commercially available asparaginase includes Colapase, Crasnitin, Crisantas, Pasum, Kidrolase, Elspar, Erwinaze PEG-asparaginase, and Pegasparagasum.

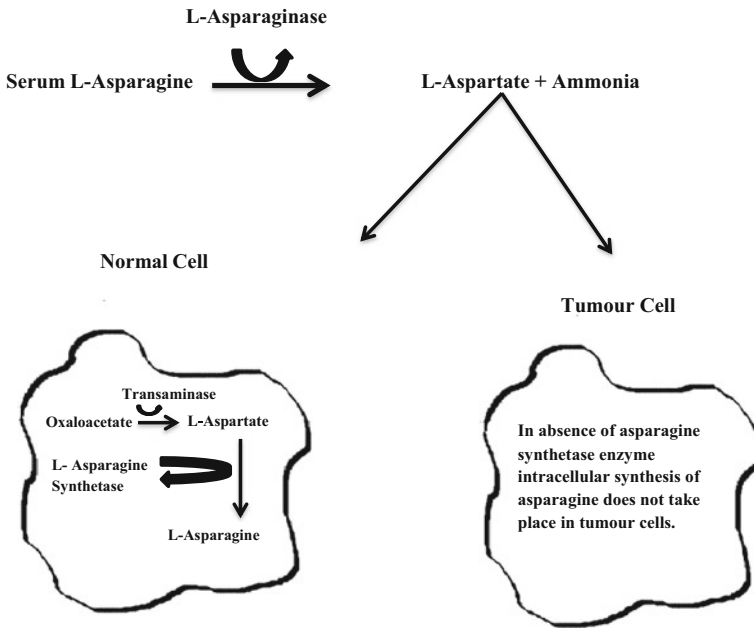
This paper reviews the different sources of L-asparaginase enzyme, methods that are being employed for its production, its assay methods, and its extensive applications in the health and food industry. Consequently, this review is organized in different sections. Section 1 sheds light on the structure, mode of action, and side effects of L-asparaginase. Section 2 gives insight into various sources of L-asparaginase enzyme including different microorganisms. Section 3 presents different production methods under submerged fermentation and solid state fermentation conditions and purification methods implemented for purification of L-asparaginase. Section 4 draws out the assay methods that are being used for detection of L-asparaginase. Section 5 discusses the applications in different areas including amino acid metabolism and clinical and food industry. Section 6 summarizes our conclusions and recommendations for future research on L-asparaginase.

### L-Asparaginase Mechanism of Action

Because of its hydrolytic property, asparaginase is a cornerstone of treatment for ALL. Both normal and leukemic cells require the amino acid L-asparagine, for their metabolic needs. Normal cells can synthesize L-asparagine for their growth by utilizing transaminase enzyme that converts oxaloacetate into an intermediate aspartate, which later on transfers an amino group from glutamate to oxaloacetate producing  $\alpha$ -ketoglutarate and aspartate. Finally, in healthy cells, aspartate is converted to asparagine by enzyme asparagine synthetase. Neoplastic cells lack the ability to synthesize the asparagine due to the absence of L-asparaginase synthetase enzyme, hence are dependent on the exogenous supply of asparagine for their existence and reproduction. Consequently, as a result, provision of L-asparaginase to tumor cells drains all circulating asparagine, which leads to starvation of cancer cells and ultimately they die off (Fig. 2). Clinical data published over the last two decades suggests asparaginase is a very important component of treatment regimens for ALL. L-asparaginase is prevalent among eukaryotes and microorganisms [20]. L-asparaginase also has a potential role in the food industry as a food processor [21].

### L-Asparaginase Structure

Research has been carried out by many scientists to elucidate the structure of L-asparaginase enzyme at the molecular level. Usually, L-asparaginase exists as a tetramer but hexameric, dimeric, and monomeric forms are also found when isolated from different sources. Most bacterial L-asparaginases exhibit quaternary and tertiary structures [22]. Molecular structures of *E. coli* and *Erwinia* sp. are very well investigated, and their structural information is easily available [23, 24]. Both *E. coli* and *Erwinia* sp. have similar three-dimensional structures [25]. *Erwinia carotovora* enzyme consists of two tetramers (ABCD and EFGH) made up of four identical monomers (A to H) each. Three hundred twenty-seven amino acids assemble in each monomer with 14  $\alpha$ -strands, eight  $\beta$ -helices [26], and two domains, a big N-terminal domain and a small C-terminal domain [24]. The active site is positioned in between two adjacent monomers (A and C: B and D). The tetramer consists of four identical subunits. The whole molecule is considered as a dimer of dimers [27, 28]. Every active site is shaped by the conveyance of amino acids in two adjacent monomers. Following amino acids constitute



**Fig. 2** Mechanism of Action of L-asparaginase in normal and tumor cells

active site: Thr15, Tyr29, Ser62, Glu63, Thr95, Asp 96, Ala120, and Lys168, while only one residue Ser254 is present in adjacent monomer [22, 27, 29–34]. Thr15 and Thr95 are the residues responsible for the catalytic activity of the enzyme.

### L-Asparaginase Side Effects

L-asparaginase causes certain side effects, and despite its potential antileukemic activity, utilization of L-asparaginase by leukemic patients causes lethality to normal cells. L-asparaginase produces a broad range of symptoms such as edema, skin rashes, fever, hepatic dysfunction, diabetes, leucopenia, pancreatitis, neurological seizures, and hemorrhage [35, 36]. Moola reported that some hypersensitivity reactions, mild allergic reactions, and anaphylactic shock are also caused by the usage of asparaginase-based drugs [37]. Adolescents appear to be at higher risk of neurotoxicity caused by L-asparaginase, which results in depression, fatigue, lethargy, dizziness, and agitation [38]. The toxicity of L-asparaginases is believed to be brought about by its glutaminase action [39].

### Sources of L-Asparaginase

The presence of L-asparaginase has been reported in various organisms, including animals, plants, and microorganisms (bacteria, fungi, algae, yeast, and actinomycetes) except humans. Even though L-asparaginase exists in several animal and plant groups, but owing to tiresome extraction procedures, other conceivable sources (bacteria, fungi, algae, yeasts, and actinomycetes) were explored by researchers. Large scale production of the enzyme from microbes is

much easier due to their facile production methods [40]. Here, we have outlined different microbial sources of L-asparaginase.

## Bacterial Sources

L-asparaginase has been reported from both Gram-positive and Gram-negative bacterial species from the terrestrial and marine environment [41]. Gram-positive bacteria have gained less consideration as compared to Gram-negative [20]. More intensively studied species of both Gram-negative and Gram-positive classes are listed in Table 1.

L-asparaginase from most of the Gram-negative bacteria can be categorized into two main types: type I and type II L-asparaginase. Type I L-asparaginase is expressed quantitatively and possesses enzymatic activity on both L-glutamine and L-asparagine amino acids, while type II L-asparaginase possesses high specific activity on L-asparagine and is only induced in anaerobic condition [80]. Type II L-asparaginase produced from *E. coli* (Ec AII) and *Erwinia*

**Table 1** Bacterial sources of L-asparaginase

Gram-negative bacteria	References	Gram-positive bacteria	References
<i>Acinetobacter calcoaceticus</i>	[42]	<i>Bacillus circulans</i>	[43, 44]
<i>Azotobacter agilis</i>	[45]	<i>B. coagulans</i>	[46]
<i>Brevibacillus brevis</i>	[47]	<i>Bacillus</i> sp.	[48]
<i>Citrobacter</i> sp.	[49]	<i>B. mesentericus</i>	[50]
<i>Escherichia coli</i>	[51]	<i>B. polymyxa</i>	[52]
<i>Enterobacter aerogenes</i>	[53]	<i>B. subtilis</i>	[54]
<i>E. cloacae</i>	[55]	<i>B. licheniformis</i>	[56, 57]
<i>Erwinia aroideae</i>	[58]	<i>B. circulans</i> MTCC 8574	[43]
<i>E. cartovora</i>	[59]	<i>Corynebacterium glutamicum</i>	[60]
<i>E. chrysanthemi</i>	[37]	<i>Mycobacterium bovis</i>	[45]
<i>Helicobacter pylori</i>	[61]	<i>M. phlei</i>	[62]
<i>Klebsiella pneumoniae</i>	[63]	<i>Staphylococcus</i> sp.	[64]
<i>Pectobacterium carotovorum</i>	[65]	<i>S. aureus</i>	[66]
<i>Pseudomonas</i> sp.	[67]	<i>Streptococcus albus</i>	[63]
<i>P. fluorescens</i> AG	[68]		
<i>P. geniculata</i>	[69]		
<i>P. ovalis</i>	[70]		
<i>P. stutzeri</i>	[71]		
<i>Pyrococcus horikoshii</i>	[72]		
<i>Serratia marcescens</i>	[73]		
<i>Thermus thermophiles</i>	[74]		
<i>T. aquaticus</i>	[75]		
<i>Vibrio succinogenes</i>	[76]		
<i>Citrobacter freundii</i>	[77]		
<i>Proteus vulgaris</i>	[78]		
<i>Zymomonas mobilis</i>	[79]		

*chrysanthemi* (Er A) has been used as an antitumor agent for the effective treatment of ALL for over 30 years [81]. L-asparaginase from two bacterial sources like *E. coli* and *E. carotovora* is currently in clinical use for the treatment of acute lymphoblastic leukemia [82].

## Fungal Sources

Fungi are another potential source of L-asparaginase along with bacteria. Several adverse side effects linked with bacterial asparaginase often restrain their application. This impediment requests a quest for L-asparaginase from new sources. Human beings are more closely related to fungi as compared to bacteria. Hence, chances of immunological reaction against fungal asparaginase will be lesser [83]. Fungal asparaginase has acquired importance based on the fact that it is produced extracellularly, and it is a very easy to purify extracellular enzyme. Several L-asparaginase-producing fungi are being reported (Table 2).

## Yeast Sources

L-asparaginase with less injurious effects is reported from the yeast. Asparaginases reported from the yeasts including *Saccharomyces* sp., *Candida* sp., *Pichia* sp., *Rhodotorula* sp., *Hansenula* sp., and *Spobolomyces* sp., exhibit antitumor activity [93]. L-asparaginase-producing yeasts are listed in Table 3.

## Actinomycete Sources

L-asparaginase occurrence is also reported in actinomycetes [100]. Actinomycetes are pervasive worldwide in soil, water, and nature, but only those found in living animals especially in fishes are found to bear good enzymatic activity [101]. Moreover, actinomycetes are a better source of L-asparaginase when compared to bacteria and fungi [102]. Table 4 lists L-asparaginase-producing actinomycetes.

## Algal Sources

Reports are available on the production of L-asparaginase from *Chlamydomonas* sp. [114] and a yellow green alga, *Vaucheria uncinata* [115].

**Table 2** Fungal sources of L-asparaginase

Fungi	References
<i>Alternaria</i> sp.	[84]
<i>Aspergillus nidulans</i>	[85]
<i>A. niger</i>	[86]
<i>A. oryzae</i>	[87]
<i>A. tamaritii</i>	[88]
<i>A. terreus</i>	[89]
<i>Cylindrocapsa obtusisporum</i>	[90]
<i>Mucor</i> sp.	[91]
<i>Fusarium roseum</i>	[92]



**Table 3** Yeast sources of L-asparaginase

Yeast	References
<i>Candida utilis</i>	[94]
<i>C. guilliermondii</i>	[95]
<i>C. bombicola</i>	[96]
<i>Pichia polymorpha</i>	[97]
<i>Rhodospiridium toruloides</i>	[98]
<i>Rhodotorula</i> sp.	[96]
<i>Saccharomyces cerevisiae</i>	[99]

## Production and Purification of L-Asparaginase

Different methods reported for production and optimization of L-asparaginase from various microorganisms include solid state fermentation (SSF) and submerged fermentation (SmF) [93]. Reaction conditions vary from one organism to another for enzyme production, and it can be produced constitutively or after induction [116, 117]. SmF process has certain constraints like net yield is very low and costly. Whereas, SSF is cost-effective and yield of product is higher as compared to the SmF. Several methods have been reported for purification of L-asparaginase enzyme produced from various sources. The major suggested requisition of L-asparaginase is as an antileukemic drug that demands a very high level of purity for this enzyme. Most of the microbial asparaginases reported so far are intracellular in nature except for the few, which are extracellular [20]. Purification of intracellular asparaginases is tiresome as compared to extracellular enzymes. Detailed reaction conditions for production and purification of L-asparaginase using SSF and SmF from different microbes have been listed in Table 5 [94].

**Table 4** L-asparaginase-producing actinomycetes

Actinomycetes	References
<i>Actinomyces</i> sp.	[82]
<i>Streptomyces albidoflavus</i>	[103]
<i>S. aurantiacus</i>	[103]
<i>S. collinus</i>	[104]
<i>S. griseus</i>	[105]
<i>S. gulbargensis</i>	[106]
<i>S. karnatakensis</i>	[107]
<i>S. longsporoflavus</i>	[108]
<i>S. plicatus</i>	[109]
<i>S. tendae</i>	[110]
<i>S. venezuelae</i>	[111]
<i>Thermoactinomyces vulgaris</i>	[112]
<i>Nocardia</i> sp.	[113]

**Table 5** Different production, purification, and optimization methods of L-asparaginase from various sources (adapted from [94])

Organism	Production Conditions		Purification		Properties			Ref.			
	Method	Medium	Temp (°C)	pH	Steps	Yield %	Purification fold		Optimum pH	Optimum temp (°C)	Mol. wt SDS-PAGE
<i>Aspergillus niger</i> AK-10	SSF	Soybean meal	–	–	–	–	–	8.6	37	91.4 kDa	[118]
<i>Bacillus</i> sp.	SmF	Glucose	37	–	Ammonium sulfate IEC DEAE	96.2 43.1	10.9 11.2	7.0	37	45 kDa	[119]
<i>Bacillus</i> sp.	SSF	Corn cob	37	5	–	–	–	5	37	–	[120]
<i>Marine actinomycetes</i> S3	SSF/SmF	Soybean meal/ tryptone glucose yeast extract broth	37	7	Ammonium sulfate Sephadex G 100	75 69.97	1.09 2	7.5	50	–	[121]
<i>Erwinia carotovora</i>	SmF	Nutrient Broth	25	6.9	A. sulfate DEAE	85 76	6 88	8.6	35	33.5 kDa	[122]
<i>Marine actinomycetes</i> PDK2	–	–	–	–	A. sulfate Sephadex G50 Sephadex G200	65.83 8.61 2.18	1.09 33.68 82.98	8.0	60	140 kDa	[123]
<i>Pseudomonas aeruginosa</i> 50071	SSF	Soybean meal	37	7.4	A. sulfate Sephadex G100 CM Sephadex C50	85 60.8 43	5.2 27.7 106	9.0	37	160 kDa	[124]
<i>Corynebacterium glutamicum</i>	SmF	Tryptone soya broth	30	7.3	Protamine sulfate DEAE Sephacel A. sulfate Sephacryl S 200	105 35 16 12.5	1.2 6.6 16.3 98	7.0	40	80±1 kDa	[60]
<i>Azotobacter vinelandii</i>	SmF	Sucrose	30	7.4	Protamine sulfate A. sulfate Sephadex G150 DEAE	90 63 21 5	– 1.6 4.5 12.5	8.6	48	84 kDa	[125]

Table 5 (continued)

Organism	Production Conditions		Temp (°C)	pH	Purification		Properties			Ref.
	Method	Medium			Steps	Yield %	Purification fold	Optimum pH	Optimum temp (°C)	
<i>Vibrio stuccinogenes</i>	–	–	–	–	Affinity chromatography	2.8	18.5	7.3	37	146 kDa [126]
<i>Streptomyces gulbargensis</i>	SmF	Groundnut cake extract, 0.5 % maltose	40	8.5	A. sulfate Sephacryl S 200 CM Sephadex C50	50.6 37.8 32	1.8 26.88 82.12	9.0	40	85 kDa [106]
<i>Pseudomonas stutzeri</i> MB-405	–	–	–	–	–	–	–	9.0	37	34 kDa [71]
<i>Fusarium equiseti</i>	SSF	Soybean meal	45	7	–	–	–	–	–	– [127]
<i>Flammulina velutipes</i>	SmF	Minimal medium	–	–	CTAB Superose 6	–	–	7.0	40	13 kDa, 85 kDa [128]
<i>Streptomyces</i> sp. TA22	SmF	Sucrose	28	7	Ammonium sulfate	–	–	7.0	–	– [129]
<i>Staphylococcus</i> s. 6A	SmF	Glucose	39	7.5	–	–	–	–	–	– [130]
<i>Thermus aquaticus</i> strain T351	–	–	–	–	DEAE-Sephacrose CL-6B column	86	18	9.5	–	80±2 kDa [75]
<i>Erwinia chrysanthemi</i> 3937 ( <i>ErL-Asparaginase</i> in	–	–	–	–	QAE-Sephadex A 50 Hydroxylapatite	83 38	29 75	–	–	–
<i>E. coli</i> BL21 (DE3) pLysS	–	–	–	–	Sephacryl G-150	21	225	–	–	–
<i>Vibrio stuccinogenes</i>	SmF	–	–	–	S-Sephacryl FF column	69.8	15.4	–	–	37.2 kDa [19]
<i>Vibrio stuccinogenes</i>	SmF	Sodium fumarate, cysteine	37	7.3–7.4	–	–	–	–	–	– [131]
<i>Escherichia coli</i>	SmF	Terrific broth	37	7.2	Ni-NTA affinity chromatography	–	3.3	–	–	37 kDa [132]

Table 5 (continued)

Organism	Production Conditions			Purification		Properties			Ref.
	Method	Medium	Temp (°C)	Steps	Purification fold	Optimum pH	Optimum temp (°C)	Mol. wt SDS-PAGE	
<i>Mycobacterium tuberculosis</i> (tiasparagine amidohydrolases)	SmF	Youman's medium	–	Ammonium sulfate	7	–	–	–	[133]
<i>Acinetobacter glutaminasificans</i> (ATCC 27197)	SmF	L-glutamic acid	–	–	–	–	–	–	[134]
<i>Staphylococcus aureus</i> strain NCTC413	SmF	Cas amino acids	–	–	–	–	–	–	[135]
<i>Fusarium tricinatum</i>	SmF	Nutrient broth with asparagine	–	A. sulfate DEAE Ampholine Prep-disk electrophoresis	16 35 22 15	8	–	161–170 kDa	[136]
<i>Acinetobacter calcoaceticus</i>	SmF	Xylose and L-asparagine	–	Precipitation with streptomycin DEAE CMC chromatography	– – 58	7–9	130 kDa	–	[42]
				Agarose filtration Phospho-cellulose Streptomycin precipitation DEAE-cellulose	14 11.6 – 3	8.4–8.8	20–30	25 kDa, 105 kDa	[137]
				Sephadex G-200 Affinity chromatography Sephacrose 6B	31 15 80	–	–	–	
<i>Serratia marcescens</i>	–	Tryptic soy broth	–	A. sulfate	80	8.5	–	–	[138]

Table 5 (continued)

Organism	Production Conditions		Purification		Properties		Ref.			
	Method	Medium	Temp (°C)	pH	Steps	Yield %		Purification fold	Optimum pH	Optimum temp (°C)
<i>Bacillus coagulans</i>	–	–	–	–	P-150 BioGel filtration	47	95.15	–	–	37 kDa, 147 kDa
	–	–	–	–	Isoelectric focusing	50	153.03	–	–	–
	–	–	–	–	Prep-disc electrophoresis	27	684	–	–	–
	–	–	37	–	DEAE	28.85	33.5	–	–	–
	–	–	–	–	BioGel P-300	13.45	130	–	–	–
<i>Bacillus coagulans</i>	–	–	–	–	Hydroxylapatite	9.5	240	–	–	–
	–	–	–	–	DEAE at pH 8.6 (11.8)	75.5	11.8	8.8–9.7	55	84 kDa
	–	–	–	–	DEAE at pH 6.6 (1.2)	55.2	1.2	–	–	–
	–	–	–	–	Sephadex G-150 (1.2)	68.5	1.8	–	–	–
	–	–	–	–	Hydroxylapatite-cellulose, pH 6.6	47.7	13	–	–	–
<i>Pseudomonas</i> sp. strain GG13	–	Mono-Na glutamate and glucose	–	–	Overall	13.6	332	–	–	25 kDa
	–	–	–	–	Butanol, A. sulfate, and zone electrophoresis	–	–	–	–	–
<i>Bacillus brevis</i>	SmF	Fructose, liquid paraffin	30	7	–	–	–	–	–	–
<i>Erwinia carotovora</i> (cloned in <i>E. coli</i> )	–	Yeast extract, soya bean meal, glucose, wood chips, asparagme, NaCl	–	–	Cation exchange and affinity chromatography	–	–	–	–	–
<i>Pectobacterium carotovorum</i> MTCC 1428	–	Glucose and L-asparagine	30	6.5	A. sulfate	70.15	6.86	8.5	40	36.5 kDa, 146 kDa
	–	–	–	–	DEAE	48.38	44.63	–	–	–
<i>Pseudomonas aeruginosa</i>	SSF	Com steep liquor, asparagine, tryptone, and casein	37	8	Sephadex G-100	42.05	72.12	–	–	–
–	–	–	–	–	–	–	–	–	–	–

**Table 5** (continued)

Organism	Production Conditions			Purification		Properties			Ref.		
	Method	Medium	Temp (°C)	Temp (°C)	Steps	Yield %	Purification fold	Optimum pH		Optimum temp (°C)	Mol. wt SDS-PAGE
<i>Escherichia coli</i> ATCC 11303	SmF	Lactose, tryptone, yeast extract, and L-asparagine	37	7.2	–	–	–	–	–	–	[143]
<i>Helicobacter pylori</i> CCUG 17874	–	–	–	–	Superdex 200 10/300 GL column	–	–	–	–	37 kDa, 140 kDa	[144]
<i>Pseudomonas</i> POT111 (cloned in <i>E. coli</i> )	SmF	Glucose, peptone, and beef extract	–	–	–	–	–	–	–	–	[145]
<i>Aspergillus niger</i>	SSF	Bran of <i>Glycine max</i>	35	6.5	A. sulfate DEAE	92.05	1.14	6.5	40	–	[87]
<i>Serratia marcescens</i>	SmF	Glycerol peptone yeast extract	37	6.8–7.0	–	–	–	–	–	–	[146]
<i>Erwinia aroideae</i> NRRL B-138	Smf	Glucose yeast extract Tryptone K <sub>2</sub> HPO <sub>4</sub>	28	7	A. sulfate	–	–	7.5	–	–	[147]

## Assay Methods

Different methods have been developed and reported to assay L-asparaginase activity. L-asparaginase activity can be evaluated by measuring the amount of ammonia or aspartic acid released during the reaction or by following the disappearance of asparagine [148, 149]. Both quantitative and qualitative techniques have been employed for estimation of L-asparaginase activity. Different methods are summarized in Table 6 [150].

## Applications of L-Asparaginase

### Anticancer Drug

L-Asparaginase is used to treat ALL in combination with vincristine and a glucocorticoid (e.g., dexamethasone) [159]. Due to its antileukemic properties, L-asparaginase has been considered as a therapeutically important antitumor drug. L-asparaginase is a well-known chemotherapeutic agent which in combination with other drugs is used in the treatment of certain malignancies such as ALL (mainly in children), Hodgkin's disease, acute myelocytic leukemia, acute myelomonocytic leukemia, chronic lymphocytic leukemia, lymphosarcoma, reticulosarcoma, and melanosarcoma [6, 7]. L-asparagine is an essential amino acid for many tumor cells for protein synthesis and cell growth, while L-asparaginase possesses the ability to convert L-asparagine to aspartate; so, in the presence of L-asparaginase, malignant cells are deprived of important growth factor that results in depletion of asparagine and ultimately tumor cells die off [160]. L-asparaginase is widely reported in animals, plants, and microorganisms, but only the asparaginase from *E. coli* and *E. chrysanthemi* was approved to be used as part of a multiagent chemotherapy to treat ALL [161]. The significant disclosure by Kidd in 1953 that guinea pig serum could stifle the development of Gardner 6C3HED lymphosarcoma cells established subcutaneously in mice opened new doors for the advancement of L-asparaginase as a treatment for ALL [160]. Table 7 [160] summarizes the breakthroughs achieved during different time periods for the development of L-asparaginase as an anticancer drug.

### Role of L-Asparaginase in Biosensor

L-asparaginase is also used for the development of a biosensor to analyze asparagine levels either in leukemia or the food industry [161]. Several spectroscopy techniques such as XRD, XPS, SEM, and TEM are currently used for L-asparagine analysis, but high cost and tedious procedures make them less favorable [162]. Over such a situation, biosensor technology can be a reliable, cheap, and user-friendly approach. The mechanism of action of the biosensor is based on asparaginase activity, ammonium ions produced from the hydrolysis of asparagine cause a change in pH resulting in the change of color and absorption [163].



**Table 6** Summary of assays for L-asparaginase activity determination [150]

No. Assay	Substrate	Product analyzed	Principle	Remarks	References
1.	L-Asparagine	Ammonia	The OD at 500 nm after the colored reaction is measured and compared to a standard curve prepared from solutions of ammonium sulfate as nitrogen source Ammonia produced is degraded by glutamate dehydrogenase with concomitant oxidation of beta-NADH; disappearance of beta-NADH is then monitored spectrophotometry at 340 nm	The method involves the use of highly toxic reagents. Simple and quantitative but low activities of L-asparaginase cannot be measured Simple and continuous assay	[147] [151]
	L-Asparagine	Ammonia	The aspartate concentration measured spectrophotometry through coupling with L-glutamic oxaloacetic transaminase and L-malic dehydrogenase	Linear results for ammonia concentrations between 0 and 200 μM	[152]
2.	L-Aspartic acid beta-(7-amido-4-methylcoumarin)	7-Amino-4-methylcoumarin	The excitation and emission wavelengths of the released 7-amino-4-methylcoumarin are measured at 37 °C using a fluorometer The ammonia released is quantitatively measured by titration	Rapid assay but such procedures require diligent attention	[153]
2.	Asparagine	Ammonia		Good reproducibility is achieved but the method requires meticulous care and disparate results are obtained from different labs analyzing the same sample	[154]
3.	Kinetic enzymatic method (gas sensing electrode)	Asparagine	The ammonia released is measured using an ammonia gas sensing electrode	Expensive but very accurate and instantaneous quantification of ammonia	[155]



Table 6 (continued)

No. Assay	Substrate	Product analyzed	Principle	Remarks	References
	Asparagine	Ammonia	The ammonia released is measured using an ammonia gas electrode (ISE)	Direct quantification of ammonia	
4.	Paper electrophoresis Asparagine	Aspartic acid	L-Aspartate was well separated from L-asparagine by electrophoresis. The section of the strip-bearing aspartate was cut out and counted in a scintillation counter	Good sensitivity but time consuming	[8, 156]
5.	Chromatography Asparagine	Aspartic acid	The conversion of L-asparagine to L-aspartic acid is followed by rapid chromatography on ion exchange paper	Detection of as small as a few pmoles of [14C] L-aspartic Acid. Therefore, a very less costly substrate required. But the method is time consuming and not continuous	[157]
6.	Conductimetry Asparagine or aspartate hydroxamate	Ammonia and/or aspartate	Increased conductivity owing to high yield of ammonia or aspartic acid in reaction mixture	Measurement is simple and instantaneous	[158]

**Table 7** Development of L-asparaginase (LA) as antileukemic medication [Adapted from 161]

Year	Advancement
1953	Kidd: discovery of antileukemic effect of guinea pig serum
1963	Broome: identification of L-asparaginase as antileukemic agent in guinea pig serum
1964–1967	Suppression of tumor cell growth by <i>Escherichia coli</i> -derived L-asparaginase; isolation and purification of active <i>E. coli</i> isoform
1966	Dolowy: first clinical use of L-asparaginase
1968	Wade: isolation of L-asparaginase from <i>Erwinia carotovora (chrysanthemi)</i>
1978	Native <i>E. coli</i> asparaginase approved by FDA for use to treat ALL
1981	Kamisaki: initial development of pegylated <i>E. coli</i> -derived asparaginase
1985	<i>Erwinia</i> asparaginase authorized in the UK for use to treat ALL
1993	Asselin: identification of distinct pharmacokinetic properties
1994	Pegylated <i>E. coli</i> asparaginase (Oncaspar) approved by FDA for use to treat ALL
2006	Pegylated <i>E. coli</i> asparaginase approved by FDA for first-line use to treat ALL
2008	Start of COG ALL07P2 and compassionate use EMTP trials to evaluate asparaginase <i>Erwinia chrysanthemi</i>
2011	Asparaginase <i>Erwinia chrysanthemi</i> approved by FDA for use in patients with hypersensitivity to <i>E. coli</i> -derived asparaginase

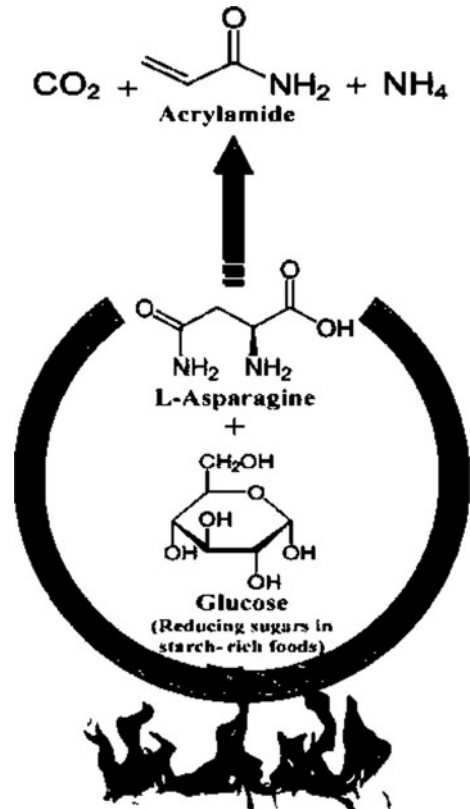
### Role of L-Asparaginase in Amino Acid Metabolism

L-Asparaginase plays a vital role in the biosynthesis of an aspartic family of amino acids, namely lysine, threonine, and methionine. Besides Krebs's cycle, aspartic acid that is a direct precursor of lysine, and threonine, is also formed by the action of L-asparaginase enzyme [164].

### Role of L-Asparaginase in Food industry

Besides, L-asparaginase is also used widely as a food processing aid. Recent developments in food technology exhibited that a colorless and odorless crystalline solid, acrylamide (2-propenamide), is produced as a result of Millard reaction, when starchy foods are fried or baked at 120 °C [165, 166]. Acrylamide is a neurotoxin and has been categorized as a carcinogenic to human [167]. In the food industry, acrylamide is largely derived from heat-induced reactions (Fig. 3) [35] between the  $\alpha$ -amino group of the free amino acid asparagine and carbonyl groups of reducing sugars such as glucose during baking and frying [168]. Due to its ability to convert L-asparagine to L-aspartate, L-asparaginase promises to be a possible way to reduce the number of precursors for Millard reaction by pre-treating the starchy foods (potato and bread dough) with L-asparaginase, hence reducing the risk of acrylamide formation [169]. However, complete removal of acrylamide is not possible due to other asparagine-independent formation [21]. Certain fungal asparaginase used in food industries nowadays includes L-asparaginase from *Aspergillus oryzae* and *Aspergillus niger* [170].

**Fig. 3** Illustration of the formation of acrylamide



## Conclusion

In this review article, we have presented both earlier and recent developments that continue to support the importance of L-asparaginase both in clinical and food industry. L-asparaginase, an amidohydrolase enzyme, is a significant enzyme that owes wide range of applications and is involved in deamination of asparagine and glutamine. From last 38 years, a lot of explorations have been carried out to establish the role of L-asparaginase in cancer therapy. Though distributed widely in bacteria, fungi, actinomycetes, plants, and animals, microbial L-asparaginase has drawn considerable attention due to cost-effective and easy production from microbes especially bacteria. However, certain limitations such as hypersensitive effects, resistance, immunogenic complications, and short life of enzyme are connected with its applications. Moreover, yields of L-asparaginase are low since gene expression rate of this enzyme is very low. Besides, L-asparaginase is utilized as food processing aid in the food industry for the elimination of carcinogenic compound acrylamide. Therefore, the desire of the time is to explore and optimize the production and purification parameters from new sources of the enzyme so that we can come up with more potential and specific L-asparaginase enzyme preparations. On the basis of better understanding of recombinant technologies like molecular cloning and genetic engineering, it may be possible to design enzyme with less immunogenic effects, novel functionalities, and better half-life. Besides, the use of latest recombinant

technologies could help to increase the net production yield of this therapeutically important enzyme.

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